

Original Article

An investigation and comparison of concentration change in simulated body fluid medium conditions of the Calcium element in 27 different *Salvia* species

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ABSTRACT

Background and Aims: In this study, calcium (Ca) element concentration changes of *Salvia* species in simulated body fluid (SBF) medium conditions were investigated and the results in these medium conditions were compared with each other.
Methods: *Salvia* species samples were air-dried and ground into powder. *Salvia* species was prepared as a herbal tea. Prepared teas were left in three different SBF medium conditions. The samples were analyzed using Flame Atomic Absorption Spectroscopy (FAAS) method to determine the Ca absorptions. Also, a correlation analysis of the results obtained in three different SBF medium conditions were compared, it was determined that the highest Ca absorption of all *Salvia* species occurred in the SGF conditions. Thus, it can be said that the Ca in plants and foods occurs in the gastric fluid medium and its absorption occurs there. When the result of the correlation analysis was evaluated, it was determined that there was a stronger correlation between the SGF and simulated intestinal fluid (SIF) mediums compared to other mediums.
Conclusion: The Ca absorption was determined according to what remained in the SGF the most. Based on the results obtained from SBF medium conditions, it can be said which element is taken in which body fluid medium. Information on the differences between samples belonging to different SBF medium conditions was not obtained. However, when the simulated fluid medium conditions was obtained individually, information was obtained for two or more samples.

Keywords: Calcium, FAAS, Salvia, SBF

INTRODUCTION

The Lamiaceae family, also known as Labiatae, is represented by more than 245 genera and 7886 species worldwide (Celep & Dirmenci, 2017). Members of this family include medicinal and aromatic species of commercial importance. These species are *Salvia* L., *Satureja* L., *Origanum* L., *Thymus* L., etc. (Kaya, Başer, Satil, & Tümen, 2000; Kurkcuoglu, Turnen, & Baser, 2001; Satıl, Ünal, & Hopa, 2007). In the flora of Turkey, this family is represented by 45 genera, 558 species, and 742 taxa (Yılar, Bayar, & Onaran, 2019).

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The genus *Salvia* belongs to the *Lamiaceae* family and is represented by more than 1000 species worldwide (Al Jaber, 2016; Gedik, Kiran, Emre, & Kursat, 2016; Nickavar, Rezaee, & Nickavar, 2016; Asadi-Samani, Khaledi, Khaledi, Samarghandian, & Gholipour, 2019; Coşge Şenkal, 2019; Güzel et al., 2019; Moridi Farimani, Miran, & Ebrahimi, 2019). This genus is distributed across Central and South America (500 species), West Asia (200 species), and East Asia (100 species) (Kahraman & Doğan, 2010; Büyükkartal, Kahraman, Çölgeçen, Doğan, & Karabacak, 2011; Kahraman, Doğan, & Celep, 2011). It is represented by 101 species in the flora of Turkey and 53 of them are endemic (Firat, 2020).

The species of this genus and the essential oils obtained from them are used in herbal tea, culinary herb, food preservatives, food flavorings, cosmetics, perfumery, and the pharmaceutical industry (Senatore, Formisano, Arnold, & Piozzi, 2005; Tosun et al., 2009; Kahraman, Celep, & Doğan, 2010; Nickavar et al., 2016; Güzel et al., 2019; Kahnamoei et al., 2019). Species of this genus are used to treat various diseases in traditional medicine in various parts of the world for wound healing, stomach ailments, alleviation of abdominal pain, liver and rheumatic pains, analgesic, antirheumatic, antioxidant, antibacterial, antimicrobial, antitumor, antidementia, cytotoxic, antiviral, carminative, diuretic, hemostatic, spasmolytic, sedative, as an antiseptic, in the treatment of hepatitis, menstrual disorders, in the treatment of colds, sedatives, stimulants, tonics, in the treatment of fever, bronchitis, tuberculosis, obesity, diabetes, depression, dementia, and menstrual disorders (Altun, Ünal, Kocagöz, & Gören, 2007; Kahraman, Celep, Doğan, & Bagherpour, 2010; Öztekin, Başkan, Kepekçi, Erim, & Topçu, 2010; Al-Qudah, Al-Jaber, Abu Zarga, & Abu Orabi, 2014; Al Jaber, 2016; Hegazy et al., 2018; Bağcı, Akbaba, Maniu, Ungureanu, & Hritcu, 2019; Bakir et al., 2020). Also, in folk medicine, Salvia species have been used since ancient times as memory-enhancing and neuroprotective agents (Çulhaoğlu, Hatipoğlu, Dönmez, & Topçu, 2015). The Salvia species in Turkey are known as "Adaçayı", "Çalba", "Şalba" and "Dağ çayı" and these species are used as herbal tea in folk medicine (Erdogan-Orhan, Baki, Şenol, & Yilmaz, 2010; Bağcı et al., 2019; Utsukarci et al., 2019). In Turkish folk medicine, Salvia species are used as antibacterials, antiseptics, diuretics, carminatives, spasmolytics, stimulants, and in treatment of wounds, colds, and coughs (Güzel et al., 2019).

For determination of element contents and concentrations of samples in the fields of chemistry, biology, medicine, pharmacy, food, environment, and agriculture various methods are used such as Atomic Absorption Spectrometry (AAS), FAAS, Graphite Furnace Atomic Absorption Spectrometry (GF-AAS), Inductively Coupled Plasma Atomic Emission Spectrometry (ICP-AES), Inductively Coupled Plasma Optical Emission Spectrometry (ICP-OES), Inductively Coupled Plasma Mass Spectrometry (ICP-OES), Inductively Coupled Plasma Mass Spectrometry (ICP-MS), Instrumental Neutron Activation Analysis (INAA), and X-ray Fluorescence Spectrometry (XRF) (Pytlakowska, Kita, Janoska, Połowniak, & Kozik, 2012; Szymczycha-Madeja, Welna, & Zyrnicki, 2013; Zhang et al., 2015; Targan, Yelboğa, & Cittan, 2018; Tunay et al., 2020).

Mineral elements are of unique and versatile importance to both plants and humans. Minerals are found in plants as ions,

inorganic and organic salts. Mineral elements are grouped into macro, micro, and ultra-micro elements. Macro-elements are generally considered as minerals of which the body needs more than 100 mg per day. These elements are C, H, O, N, P, K, Ca, Mg, Na, and S (Umaz, 2021). Ca is responsible for bone formation and metabolism, vascular contraction and vasodilation, muscle function, nerve conduction, intracellular signaling, and hormonal secretion. It also provides strength to bones and teeth with phosphate (Ross, Taylor, Yaktine, & Del Valle, 2011; Lippert, 2020; Shkembi & Huppertz, 2022).

Ca is absorbed in the ionized form (Ca²⁺) in the gastrointestinal tract (a long tubular structure between the mouth and anus and a system that includes many organs associated with this structure and whose main task is digestion) (Shkembi & Huppertz, 2022). Ca is absorbed by active transport (intercellular) and passive diffusion (extracellular) across the intestinal mucosa. Active transport of the Ca is dependent on the action of calcitriol and the intestinal vitamin D receptor (VDR). Passive diffusion involves the movement of calcium between mucosal cells. this diffusion is dependent on luminal: serosal electrochemical gradients (Ross et al., 2011).

This study has the distinction of being the first in terms of examining where the Ca element in the species is absorbed under body fluid medium conditions and the concentration change. In this study, 27 different *Salvia* species used for herbal tea, medicinal, and various purposes were prepared as herbal tea and left in three different SBF mediums. The concentration variation of the Ca element, which is important for plants and humans, was investigated and compared in SBF medium conditions. Correlation analysis between the species of Ca in SGF medium conditions was performed. In addition, in order to determine whether the correlation coefficients between the obtained results were the same or not, the correlation coefficients on the basis of Ca in different SBF medium conditions were calculated and compared.

MATERIALS AND METHODS

Herbal materials

Herbarium samples of 27 *Salvia* species collected from different parts of Turkey were prepared to be placed in the herbarium and dried in accordance with the purpose of the study. Dried species were identified and preserved by Mehmet Firat to be placed in Van Yüzüncü Yıl University, Faculty of Science Herbarium (VANF) (Table 1).

Chemicals and reagents

In the study, analytical grade chemicals were used for the preparation of three different SBF medium conditions. NaCl (EMSURE, for Analysis), MgCl₂.6H₂O (for Analysis), CH₃COOK (EMSURE, for Analysis), Lactic acid (EMPROVE), NaBr (EMPROVE), CaCl₂ (EMSURE ACS reagent), Pepsin (Biochemistry) for Analysis), KH₂PO₄ (for EMSURE Analysis), D(+)-Fructose (for Biochemistry), and NH₄OH (EMSURE for Analysis) were purchased from Merck (Germany). D(+)-Glucose (anhydrous), NaOH (ACS reactive), Urea (ACS reactive), HCl (\geq 37% ACS reagent) was purchased from Sigma Aldrich (Germany). CaCl₂.2H₂O (ACS reagent), H₃PO₄ (\geq

Plant species	Species Codes	Gathering Places	Harvest- ing Times	Herbarium Number
Salvia blepharochlaena Hedge & HubMor.	S1	Nevşehir	2014	M. FIRAT 32102 (VANF)
Salvia brachyantha (Bordz.) Pobed. subsp. brachyantha	S2	Van	2015	M. FIRAT 32469 (VANF)
Salvia candidissima Vahl subsp. candidissima	S3	Van-Gürpınar	2015	M. FIRAT 32092 (VANF)
Salvia ceratophylla L.	S4	Nevşehir	2015	M. FIRAT 32174 (VANF)
Salvia cerino-pruinosa Rech.f. var. cerino-pruinosa	S5	Elazığ	2015	M. FIRAT 32539 (VANF)
<i>Salvia cerino-pruinosa</i> Rech.f. var. <i>elazigensis</i> Kahraman, F.Celep & Dogan	S6	Elazığ	2015	M. FIRAT 32539 (VANF)
Salvia divaricata Montbret & Aucher ex Benth.	S7	Erzincan-Kemah	2017	M. FIRAT 33829 (VANF)
<i>Salvia hydrangea</i> DC. ex Benth.	S8	Kars-Kagızman	2014	M. FIRAT 30696 (VANF)
Salvia hypargeia Fisch. & C.A.Mey.	S9	Van	2015	M. FIRAT 31173 (VANF)
Salvia indica L.	S10	Hakkari	2015	M. FIRAT 32455 (VANF)
Salvia kronenburgii Rech. f.	S11	Van-Gürpınar	2014	M. FIRAT 30650 (VANF)
Salvia kurdica Boiss. & Hohen. ex Benth.	S12	Şırnak	2014	M. FIRAT 32614 (VANF)
Salvia limbata C.A.Mey.	S13	Van	2014	M. FIRAT 30660 (VANF)
Salvia macrochlamys Boiss. & Kotschy	S14	Van	2014	M. FIRAT 30907 (VANF)
Salvia microstegia Boiss. & Balansa	S15	Hakkari	2015	M. FIRAT 32472 (VANF)
Salvia montbretii Benth.	S16	Diyarbakır	2015	M. FIRAT 32463 (VANF)
Salvia multicaulis Vahl.	S17	Van	2014	M. FIRAT 30656 (VANF)
Salvia pachystachys Trautv.	S18	Van	2015	M. FIRAT 30878 (VANF)
Salvia pinnata L.	S19	Diyarbakır	2014	M. FIRAT 31318 (VANF)
Salvia pseudeuphratica Rech.f.	S20	Elazığ	2015	M. FIRAT 32584 (VANF)
Salvia sclarea L.	S21	Van-Bahçesaray	2014	M. FIRAT 30921 (VANF)
<i>Salvia siirtica</i> Kahraman, Celep & Doğan	S22	Hakkari	2014	M. FIRAT 30755 (VANF)
Salvia spinosa L.	S23	Mardin	2016	M. FIRAT 30908 (VANF)
Salvia suffruticosa Montbret & Aucher ex Benth.	S24	Van	2014	M. FIRAT 30657 (VANF)
Salvia trichoclada Benth.	S25	Van-Çatak	2014	M. FIRAT 30658 (VANF)
Salvia viridis L.	S26	Adana	2015	M. FIRAT 32461 (VANF)
Salvia xanthocheila Boiss. ex Benth.	S27	Van	2014	M. FIRAT 30668 (VANF)

Table 1. Herbarium number, collection places, and collection times of plants belonging to 27 different Salvia
species.

99%), NaF (98.5-100.5%) were purchased from Honeywell/Fluka (USA). Uric acid (\geq 99% for Biochemistry, Roth Germany), and K₃PO₄.3H₂O (97%) were purchased from abcr Gmbh (Germany). The Ca standard (1000 mg/L, Plasma CAL Calibration solution SCP28AES) in FAAS measurements was used. Linearity was evaluated using the least 6-point 3-parallel matrix calibration. Calibration curves were determined with six concentrations (5, 10, 25, 50, 75, and 100 mg/L). Calibration standard solutions (5-100 mg/L) were prepared by appropriate dilution of stock Ca standard (1000 mg/L). LOD and LOQ values of the Ca element were calculated using 10 independent blank solutions.

Preparation of SBF Preparation of simulated saliva fluid (SSF)

1.28 g NaCl was put into 1 L flask and dissolved in 500 mL ultrapure water. Then, 0.125 g MgCl.6H₂O, 0.095 g KCl, 1.508 g CH₃COOK, 0.167 g CaCl₂, 0.386 g K₃PO₄.3H₂O, 0.0042 g NaF, and

0.05 mL H_3PO_4 were added to the solution and mixed. The residue and turbidity formed in the solution were clarified by adding 1 drop of H_3PO_4 . Then, ultrapure water was added and mixed so that the final volume of the solution was 1 L. The solution was mixed by adding lactic acid according to the desired pH. The pH of the solution was adjusted to 6.3 (Shannon, 1982).

Preparation of SGF

0.265 g CaCl₂.2H₂O was put into a 1 L flask and dissolved in 500 mL of ultrapure water. Then, the solution was mixed by adding 0.153 g MgCl.6H₂O, 0.865 g KCl, 2.856 g NaCl, 0.0008 g NaBr, 0.0009 g NaF, 0.0003 g CuCl₂.2H₂O, 0.138 g D(+)-fructose, 0.350 g D(+)-glucose, 0.084 g Urea, 0.0084 g uric acid and 3.20 g pepsin. Then, ultrapure water was added and mixed so that the final volume of the solution was 1 L. The solution was mixed by adding 0.04 M HCl and 0.1 M NH₄OH according to the desired pH. The pH of the solution was adjusted to 1.54 (Stefaniak et. al., 2010).

Preparation of SIF

 $6.80 \text{ g KH}_2\text{PO}_4$ was put into a 1 L flask and dissolved in 500 mL of ultrapure water. Then, 0.90 g of NaOH was added to the solution and mixed. Then, ultrapure water was added and mixed so that the final volume of the solution was 1 L. The solution was mixed by adding 2 M HCl according to the desired pH. The pH of the solution was adjusted to 6.80 (Stippler et. al., 2004).

Preparation of Salvia species as herbal tea

0.5 g *Salvia* species was put into a 100 mL beaker and washed two times with ultrapure water to remove dust and residues. Then, 50 mL boiled ultrapure water was added to the beaker and was left to brew for 10 min. Then, the tea in the beaker was filtered with blue banded filter paper. The resulting filtrate was used in experimental studies.

Experimental designs in SBF medium of prepared herbal tea

50 mL SBF was placed in a 250 mL beaker and 50 mL of prepared herbal tea was added to it. The mixture was put on a mixer and mixed to rotate at 50 rpm. The stirring times designed for the simulated mediums were as follows: 30 seconds for SSF medium, 180 minutes for SGF medium, and 300 minutes for SIF medium. Then the mixture was filtered with filter paper and the samples were placed in screw-capped tubes to be read in the FAAS device. The above-mentioned procedures were performed separately for each SBF medium. Analysis was performed in three replicates for simulated saliva, gastric and intestinal fluid medium. In the study, 27 *Salvia* species were used and three different SBF mediums were studied based on the duration of food intake.

Table 2. FAAS instrument of conditions	operating analytical
Measured Element	Ca
Wavelength	422.7 nm
Slit Width	0.7 nm
Lamp Current	10 mA
Gas Flow Rate	2 L/min
Flame Height	7 mm
Flame Type	Air-Acetylene

FAAS analysis

The Ca analysis in SBF medium of species samples and prepared herbal tea was performed using FAAS (AA-7000, Shimadzu, Japan) (Table 2).

The linear range, calibration equation, correlation coefficient (R²), limit of detection (LOD), and limit of measurement (LOQ) values are shown in Table 3. The sensitivity of the method was evaluated by determining the limits of detection (LOD) and limits of quantification (LOQ). The LOD and LOQ values were determined under FAAS conditions from 3S/m and 10S/m respectively. The S was the standard deviation of the blank and m was the slope of the calibration equation. The LOD and LOQ values in SSF, SGF, and SIF medium conditions were found as 1.440 and 4.751 mg/L; 1.309 and 4.320 mg/L; 2.524 and 8.328 mg/L, respectively. The correlation coefficients (R²) in SSF, SGF, and SIF medium conditions were determined as 0.990, 0.995, and 0.998, respectively (Table 3).

Statistical analysis

The analysis of the species was replicated three times and the mean values of the data were used in the statistical analysis. The correlation of Ca elements in 27 *Salvia* species structures between three different SBF medium conditions was determined using linear correlation with SSPS 21 statistical package program (2012).

RESULTS AND DISCUSSION

Variation in SBF medium conditions of Ca concentration Ca is known to be an important mineral element for plants and humans. For this reason, regarding the Ca element in the species which is used for medicinal and various purposes, it is important to know what ratio is absorbed especially in humans. This study was carried out to estimate the amount that can be taken up by the organism of an element found in the digestible materials (herbal tea, food, and medicines). When the Ca absorption in SBF medium conditions of *Salvia* species was examined, it was determined that the Ca absorption of the species varied between 118-373 mg/kg in SSF medium conditions. The Ca absorption of the S22 sample was determined to be higher than other species in SSF medium conditions. In addition, the Ca absorptions of almost all species were detected

		SSF			
Element	Linear Range (mg/L)	Calibration Equation	R ²	LOD (mg/L)	LOQ (mg/L)
Ca	25-100	y=0.0028104x - 0.19540	0.990	1.440	4.751
		SGF			
Element	Linear Range (mg/L)	Calibration Equation	R ²	LOD (mg/L)	LOQ (mg/L)
Ca	5-75	y= 0.0038165x - 0.25492	0.995	1.309	4.320
		SIF			
Element	Linear Range (mg/L)	Calibration Equation	R ²	LOD (mg/L)	LOQ (mg/L)
Са	10-100	y= 0.0025610x - 0.20674	0.998	2.524	8.328

to be close to each other. As a result, it was determined that the Ca element was taken into the SSF medium in all 27 *Salvia* species (Table 4). When the Ca absorption between the species is compared under SSF medium conditions, it can be said that higher absorption in the S22 species gives more Ca²⁺ ions to the medium.

	SSF	SGF	SIF
Samples	Ca (mg/kg)	Ca (mg/kg)	Ca (mg/kg)
S1	230±27	18300±421	11096±844
S2	263±22	27511±109	13119±123
S3	308±19	33740±344	11352±164
S4	237±12	31838±810	11322±219
S5	275±4	37184±149	9398±65
S6	213±21	37723±166	11644±95
S7	195±22	31521±566	11256±633
S8	294±16	39783±151	12287±128
S9	214±46	37445±501	11494±678
S10	288±19	42316±184	12290±431
S11	267±19	27030±787	9307±304
S12	271±6	36991±113	9521±272
S13	296±20	40034±120	10182±438
S14	147±13	39442±513	10012±650
S15	283±18	38128±470	10302±755
S16	211±19	62374±159	10023±465
S17	187±14	55063±241	8415±112
S18	251±24	38177±150	8715±169
S19	330±17	38881±411	9104±120
S20	219±17	37626±405	8904±207
S21	263±21	37981±817	8976±490
S22	373±10	37746±766	8521±223
S23	292±19	36990±247	8521±406
S24	118±7	37668±626	9307±172
S25	210±14	37210±672	9458±209
S26	240±6	37131±158	12628±272
S27	262±6	38923±268	9120±82

The Ca absorption in the species was determined that varied between 18300-62374 mg/kg in SGF medium conditions. The Ca absorption of the S16 sample was determined to be higher than the other species in SGF medium conditions. It was determined that in all *Salvia* species, the Ca was taken into the SGF medium (Table 4).

The Ca absorption of the species was determined as varying between 8415-13119 mg/kg in SIF medium conditions. The Ca absorption of the S2 sample was determined to be higher than the other species in SIF medium conditions. It was determined that the Ca was taken into SIF medium in all *Salvia* species (Table 4).

Schwedt, Tawali, & Koch, (1998) the total content of zinc in the extraction with simulated gastric and intestinal fluid of foodstuffs has beendetermined. In simulated gastric juice, the total zinc content in Buckwheat flour, Rye flour, Potato, Pea, and Beef products was reported as 3.628, 1.222, 0.362, 0.751, and 4.654 mg/kg, respectively. In simulated intestinal juice, the total zinc content in the products was determined as 1.023, 0.130, 0.256, 0.371, and 3.562 mg/kg, respectively. It has been stated that the zinc element is mostly absorbed from the gastric juice (Schwedt et. al., 1998).

Giacomino et. al., (2014) the metal bioaccessibility of three drugs purchased in India has been investigated by extraction with solutions simulating gastric and intestinal fluids. The concentrations of the elements extracted upon contact between gastric and intestinal juices of the samples of drug C, D and F were reported as 1740 and 2170, 7640 and 5091, 1893 and 2490 mg/kg, respectively. Generally, it has been specified that the release of most of the analytes in the gastric medium was lower than that in the intestinal one (Giacomino et. al., 2014).

Wang et. al., (2019) the uptake of Fe, Zn, Ca, and Mg in gastric and intestinal fluids of acidic heteropolysaccharide (LP) from *Lycium barbarum* L. leaves have been determined. The Ca intakes in the gastric and intestinal fluids of the crude LP (LPC) have been reported as 0.19 and 0.54 mg, respectively. It has been stated that the Ca element is mostly absorbed from the intestinal juice (Wang et. al., 2019).

When Table 4 was examined, the ranking of Ca absorption of Salvia species in SBF medium conditions was determined as SGF > SIF > SSF. Thus, it was determined that the Ca element was absorbed most in gastric fluid and least in saliva fluid (Table 4). It can be said that the absorption of Ca in the saliva fluid being less than in other body fluids is due to the fact that the pH of the saliva is close to neutral. It has been reported in the literature that saliva has only a negligible effect on the level of mobilization of metal contaminants, as the pH of saliva is close to neutral (Giacomino et. al., 2014). When the SGF and SIF medium conditions were compared, the Ca absorption of Salvia species was determined as being higher in SGF medium conditions. It can be said that the most absorption of Ca in the gastric fluid medium is due to the acid pH of the gastric juice permiting a greater dissolution of the hydroxides, oxides, other salts, and organic species of the elements contained in the plant. Although the pH of the intestinal fluid was close to neutral, the Ca absorption in the species was determined as being high. Thus, it reveals that the Ca found in food, medicine, and herbal teas is absorbed in the intestine. Consequently, the Ca element, its carbonates, oxides, sulfites, and salts in plants and foods are mostly absorbed in gastric and intestinal fluid. It can be said that in human and animal bodies, Ca passes through the gastric and intestinal medium. When the Ca is evaluated in general, it can be said that it is absorbed in the gastric medium and is taken into the living body. The findings of this study showed that the Ca element in Salvia species was absorbed from the SGF and SIF fluid medium and was similar to the results in the literature.

Correlation analysis of SGF

It was aimed to determine whether there were differences between species according to SGF of the correlation coefficient and direction with correlation analysis. Correlation analysis for species in SGF medium conditions was used to provide information in determining whether there is a relationship in terms of Ca absorption between species that are related to each other and the relationship between species in the same region or close locations. Therefore, correlation analyses were performed for species in SGF medium conditions (Table 5 and 6). According to the results of the bivariate analysis of the samples belonging to SGF, the Pearson correlation coefficient (r) was at p < 0.01 confidence level; it was determined that there was a strong negative relationship in terms of Ca absorption between S5 with S4, S16 with S7, and S19 with S8 (r = -0.999). In addition, it was determined that there was a strong positive relationship (r = 0.999) in terms of Ca absorption between S26 with S21 species in the same medium conditions. When the Pearson correlation coefficients of the samples belonging to these simulated medium conditions were compared, it was determined that the strongest correlation at p < 0.01 confidence level was between S5 with S4, S16 with S7, S19 with S8, and S26 with S21 (Table 5).

When the p < 0.05 confidence level of the Pearson correlation coefficient (r) of the samples belonging to SGF was examined, it was determined that there was a strong negative relationship in terms of Ca absorption between S12 with S2, S12 with S9, S15 with S11, S17 with S12, S17 with S16, S23 with S20 and S24 with S3 (r = -0.999). It was determined that there was a strong negative relationship in terms of Ca absorption between S23 with S1 (r = -0.998), S26 with S9, and S19 with S11 (r = -0.997). In addition, it was determined that there was a strong positive relationship in terms of Ca absorption between S9 with S2, S17 with S7, S25 with S6 (r = 0.999), S17 with S2, S27 with S6, S11 with S8, S15 with S13 (r = 0.998) (Table 5).

Davis has been examined the close kinship of *Salvia* species with each other. He assigned a number to each *Salvia* species. According to these numbers, he said that the species between 1-7, 8-14, 15-31, 32-41, 42-47, 48-83, and 84-86 showed close kinship relationships (Table 6) (Davis, 1982). The relatedness of *Salvia* species to each other is defined by the closeness to each other of the numbers given to them. In this context, when the correlation coefficient and effect were examined, according to the results of the correlation analysis at p < 0.01 confidence level, a strong relationship between the species in terms of calcium absorption was determined. However, it was determined that there is no relationship between consanguinity and Ca absorption within these species (Table 6).

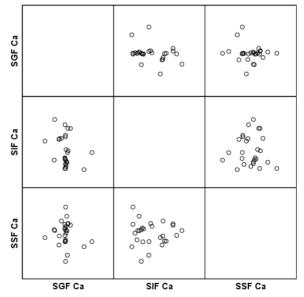
At the p < 0.05 confidence level, it was determined that there was a relationship in terms of calcium absorption between S26 (Species Kinship Code: 43) with S9 (44), S11 (38) with S8 (36), and S15 (57) with S.13 (71) which only have kinship to each other from the species which has a strong relationship with each other. As a result, it was determined that although there was a strong relationship between species in terms of calcium absorption at both p < 0.01 and p < 0.05 confidence levels, there was generally no relationship between the closely related species with each other (Table 6).

When the correlation coefficient and relationship between species collected from in the same region were evaluated, it was determined that there was only a relationship in terms of Ca absorption between S15 with S11, S24 with S3, S9 with S2. S17 with S7, S17 with S2, S27 with S6, S11 with S8, and S15 with S13 (Eastern Anatolia Region) from the species with a strong relationship with each other in the p < 0.05 confidence level. Dogan et. al., (2008) Salvia made the intra-genus classification. They divided the species into seven different sections. These sections are Hymenosphace Benth., Aethiopis Benth., Plethiosphace Benth., Horminum Benth., Drymosphace Benth., Hemisphace Benth., and Salvia Hedge. They grouped the species based on these sections (Table 6) (Dogan et. al., 2008). When the results of correlation analysis at p < 0.01 confidence level were examined at the intra-genus classification level, it was determined that there was no relationship between species in terms of Ca absorption. At the p < 0.05 confidence level, it was determined that there was a relationship in terms of Ca absorption only between S8 with S11 (Hymenosphace-Hymenosphace), S13 with S15 (Aethiopis-Aethiopis), and S9 with S2 (Aethiopis-Aethiopis) from the species with a relationship with each other (Table 6). Thus, the significant correlation between Ca element concentrations in species indicates a similar ability or presence of the same source for the Salvia species. The dependence between the species with the strongest correlation above can be explained by their common origin.

Correlation analysis of simulated body fluid medium conditions

It was aimed to determine whether the correlation coefficient and its direction differ in three different SBF medium conditions. For this reason, correlation analysis were performed for three different SBF medium conditions (Table 7) (Figure 1).

According to the correlation matrix of the Ca element in three different SBF medium conditions, negative correlations were detected between SGF with SIF (r=-0.276), SGF with SSF (r=-0.132)





S3 S2	S1	S2	S	S4	S5	Ş	S7	85	6S	S10	S11	S12	S13	S14 S	S15 S1	S16 S	S17 S18	S19	<u>S</u> 20	<u>S</u> 21	<u>S22</u>	S23 9	S24 S	S25 S26	6 S27
	-																								
	0,725	-																							
	0,988	0,822	-																						
S4 0	0,977	0,561	0,932	-																					
S5 -0	-0,978	-0,566	-0,935	-0,999	-																				
56	-0,925	-0,408	-0,855	-0,985	0,984	-																			
S7 0	0,794	0,994	0,878	0,646	-0,650	-0,503	-																		
SB 0	0,900	0,953	0,956	0,786	-0,790	-0,667	0,98	-																	
S9 0	0,702	0 ,999 ⊍	0,803	0,533	-0,538	-0,378	0,99	0,942	-																
S10 -0	-0,855	-0,262	-0,765	-0,946	0,944	0,988	-0,363	-0,543	-0,230	-															
S11 0	0,926	0,932	0,973	0,823	-0,827	-0,712	0,965	₀ 3 98₀	0,919	-0,595	-														
S12 -0	-0,739	-0'666'0	-0,834	-0,578	0,583	0,427	-0,996	-0,959	-0 ' 999	0,282	-0,939	-													
S13 -0	-0,961	-0,886	-0,992	-0,880	0,883	0,784	-0, 931	-0,985	-0,871	0,679	-0,994	0,896	-												
S14 0	0,254	-0,482	0,103	0,455	-0,450	-0,603	-0,386	-0,193	-0,511	-0,719	-0,131	0,463	0,022	-											
S15 -0	-0,944	-0,912	-0,983	-0,852	0,855	0,747	-0,950	-0,993	-0,898	0,635	-0' 999	0,920	°866′0	0,079	-										
S16 -0	-0,793	-0,994	-0,877	-0,644	0,649	0,502	-0'666"	-0,979	-0,990	0,361	-0,965	0,996	0,930	0,388 0,9	0,950 1	_									
S17 0	0,769	0 ,998 ⁵	0,858	0,615	-0,619	-0,468	0 , 999 ⁵	0,971	0,995	-0,325	0,954	-0 ' 999 ⁶	-0,915	-0,423 -0,	-0,937 -0,9	-0,999	-								
S18 0	0,175	0,805	0,324	-0,040	0,034	0,213	0,737	0,586	0,824	0,362	0,534	-0,792	-0,439	-0,908 -0,	-0,490 -0,7	-0,739 0,	0,764 1								
S19 -0	-0,895	-0,956	-0,953	-0,779	0,782	0,658	-0,982	-0,999	-0,946	0,533	-0 ' 667 b	0,962	0,983	0,204 0,9	0,992 0,9	0,982 -0,	-0,973 -0,596	1							
S20 0	0,995	0,652	0,968	0,993	-0,994	-0,958	0,729	0,852	0,627	-0,902	0,883	-0,668	-0,929	0,349 -0,	-0,906 -0,7	-0,728 0,	0,701 0,076	-0,846	٢						
S21 -0	-0,638	-0,993	-0,748	-0,458	0,463	0,297	-0,975	-0,910	-0,996	0,145	-0,882	066'0	0,825	0,583 0,8	0,856 0,9	0,975 -0,	-0,983 -0,870	0,914	-0,558						
522 -0	-0, 705	-0,023	-0,588	-0,840	0,837	0,922	-0,129	-0,326	0,011	0,971	-0,385	0,044	0,483	-0,865 0,4	0,431 0,1	0,127 -0,	-0,089 0,575	0,315	-0,772	-0,097	-				
S23 -0	-0 , 998 ⁵	-0,679	-0,976	-0,989	0,989	0,947	-0, 753	-0,870	-0,654	0,886	-0,900	0,694	0,942	-0,316 0,9	0,921 0,7	0,752 -0,	-0,726 -0,111	0,864	- 0'999	0,587	0,749	-			
524 -0	-0,991	-0,811	-0,999	-0,939	0,941	0,865	-0,868	-0,950	-0,791	0,777	-0,968	0,823	0,990	-0,122 0,9	0,980 0,8	0,868 -0,	-0,848 -0,305	5 0,947	-0,973	0,735	0,604 (0,980	-		
S25 -(-0,916	-0,388	-0,844	-0,981	0,979	0,999⊍	-0,484	-0,650	-0,357	0,991	-0,696	0,407	0,770	-0,621 0,	0,732 0,4	0,482 -0,	-0,448 0,234	0,641	-0,952	0,275	0,930 (0,940 0	0,854		
S26 -0	-0,647	-0,994	-0,756	-0,469	0,474	0,308	-0,977	-0,915	-0 ,997 b	0,157	-0,887	0,992	0,832	0,573 0,8	0,862 0,9	0,978 -0,	-0,985 -0,864	t 0,919	-0,568	°666'0	-0,080	0,597 0	0,743 0,	0,287 1	
527 -0	-0,948	-0,468	-0,888	-0,994	0,993	0,998⊍	-0,559	-0,715	-0,438	0,975	-0,757	0,487	0,824	-0,549 0,	0,790 0,5	0,558 -0,	-0,526 0,147	0,706	-0,975	0,359	0,894 (0,967 0	0,897 0,	0,996 0,371	1 1

Species	Species Codes	Species Kinship Code	Sections
Salvia blepharochlaena	S1	34	Hymenosphace Benth.,
Salvia brachyantha subsp. brachyantha	S2	49	Aethiopis Benth.,
Salvia candidissima subsp. candidissima	S3	66	Aethiopis Benth.,
Salvia ceratophylla L.	S4	53	Aethiopis Benth.,
Salvia cerino-pruinosa var. cerino-pruinosa	S5	-	Hymenosphace Benth.,
Salvia cerino-pruinosa var. elazigensis	S6	-	Hymenosphace Benth.,
Salvia divaricata	S7	1	Salvia Hedge (Eusphace Benth.,
Salvia hydrangea	S8	36	Hymenosphace Benth.,
Salvia hypargeia	S9	44	Aethiopis Benth.,
Salvia indica L.	S10	72	Aethiopis Benth.,
Salvia kronenburgii	S11	38	Hymenosphace Benth.,
Salvia kurdica	S12	6	Salvia Hedge (Eusphace Benth.,
Salvia limbata	S13	71	Aethiopis Benth.,
Salvia macrochlamys	S14	7	Salvia Hedge (Eusphace Benth.,
Salvia microstegia	S15	57	Aethiopis Benth.,
Salvia montbretii	S16	45	Aethiopis Benth.,
Salvia multicaulis	S17	40	Hymenosphace Benth.,
Salvia pachystachys	S18	27	Salvia Hedge (Eusphace Benth.,
Salvia pinnata L.	S19	11	Salvia Hedge (Eusphace Benth.,
Salvia pseudeuphratica	S20	-	Hymenosphace Benth.,
Salvia sclarea L.	S21	50	Aethiopis Benth.,
Salvia siirtica	S22	-	-
Salvia spinosa L.	S23	46	Aethiopis Benth.,
Salvia suffruticosa	S24	29	Salvia Hedge (Eusphace Benth.,
Salvia trichoclada	S25	14	Salvia Hedge (Eusphace Benth.,
Salvia viridis L.	S26	43	Horminum Benth.,
Salvia xanthocheila	S27	58	Aethiopis Benth.,

Species kinship between 1-7, 8-14, 15-31, 32-41, 42-47, 48-83 and 84-86.

	orrelation matrix medium conditi		simulated
	SGF	SIF	SSF
SGF	1		
SIF	-0.276	1	
SSF	-0.132	-0.027	1

and SSF with SIF (r= -0.027). When the correlations in SBF medium conditions were evaluated together, they were determined that SGF-SIF > SGF-SSF > SSF-SIF. When all the correlation coefficients were evaluated in general, it was determined that all SBF medium conditions were weakly correlated with each other. In addition, considering all correlation coefficients, SGF and SIF medium were detected to have stronger negative correlations than other mediums. The fact that these correlations were different from the specified SBF mediums can be explained as the result of taking other elements such as the Ca element into the medium.

CONCLUSION

Plants have medical benefits because they contain various bioactive compounds as well as mineral elements. For this reason, it is important to know the contents of mineral elements in plants and how much these elements are absorbed. In this context, this study has determined the Ca content and absorption of 27 Salvia species in three different SBF conditions. In addition, the correlation analysis of the obtained results has been performed.

According to the analysis results in different SBF medium conditions of the samples belonging to Salvia species, it has been determined that the Ca absorption in the SGF the most. Based on the results obtained from SBF medium conditions, it can be said which element is taken in which body fluid medium. In addition, it can be said that element-based synthesized drugs are a guide in the selection of the target region they want to affect.

According to the results of the bivariate analysis in SBF medium conditions, information on the differences between the samples belonging to different SBF medium conditions could not be obtained. However, when SBF medium conditions are evaluated alone, the only information that may indicate the same source or behavior is obtained for two or more items. Thus, it was determined that more complex analysis systems should be used to determine the difference in terms of the medium conditions.

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